Chemistry 1	Name
ALT5 Packet #3 Thermochemistry	Period Date
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This packet will be scored using the Packet Rubric and con	tributes to your academic grade for the unit

(1) Please read the ALT 5 learning targets and then circle the level of your current understanding:

AST 5.4 Develop a model to illustrate that the release or absorption of energy from a chemical reaction system depends upon the changes in total bond energy.

4—I totally get it, 3 - I get it. 2 - I'm starting 1 - Not getting it yet. I could teach it.

(2) Please mark one check box for each assignment and tally the marks; **Due Wed., 6/8—Test Day.**

Assignment/Worksheet/Lab	Not	D 41 11	
	Present	Partially Complete	100% Complete
Intro Endothermic & Exothermic Reactions			
2. Text p 480 (#1,2, 5)			
3. Fundamentals of Exp. Design –booklet			
4. Lesson 3 Point of View #1-6 completed			
5. Quiz #1 Endo/Exo Reactions			
6. Bond Energy Guided Inquiry			
7. Quiz #2 Bond Energy			
8. Text p. 511 (#2 and 4)			
9. Text p 515 (#1 and 2)			
10. Review Quiz			
A: COLUMN TOTALS (Tally the checks in each column)			
B. COLUMN PERCENT OF TOTAL			
	2. Text p 480 (#1,2,5) 3. Fundamentals of Exp. Design –booklet 4. Lesson 3 Point of View #1-6 completed 5. Quiz #1 Endo/Exo Reactions 6. Bond Energy Guided Inquiry 7. Quiz #2 Bond Energy 8. Text p. 511 (#2 and 4) 9. Text p 515 (#1 and 2) 10. Review Quiz A: COLUMN TOTALS (Tally the checks in each column)	2. Text p 480 (#1,2, 5) 3. Fundamentals of Exp. Design –booklet 4. Lesson 3 Point of View #1-6 completed 5. Quiz #1 Endo/Exo Reactions 6. Bond Energy Guided Inquiry 7. Quiz #2 Bond Energy 8. Text p. 511 (#2 and 4) 9. Text p 515 (#1 and 2) 10. Review Quiz A: COLUMN TOTALS (Tally the checks in each column) B. COLUMN PERCENT OF TOTAL (# of checks in box ÷10) x 100	2. Text p 480 (#1,2, 5) 3. Fundamentals of Exp. Design –booklet 4. Lesson 3 Point of View #1-6 completed 5. Quiz #1 Endo/Exo Reactions 6. Bond Energy Guided Inquiry 7. Quiz #2 Bond Energy 8. Text p. 511 (#2 and 4) 9. Text p 515 (#1 and 2) 10. Review Quiz A: COLUMN TOTALS (Tally the checks in each column) B. COLUMN PERCENT OF TOTAL (# of checks in box ÷10) x 100

Chemistry 1 Coursework Grading Rubric

Teacher will select 6-10 problems from handouts and text book problems to be graded for correctness.

Exam Date: 6/8/16

4	☐ Packet is missing 1 or no assignments (only 1 checks in near left column)
	All the handouts/problems completed (all checks in far right column)
	☐ Showed all work and not just answers.
	☐ Answers to circled graded problems are correct/accurate.
3	☐ Packet is missing 2-3 items (2 or 3 checks in leftmost column).
	3 or fewer partially complete assignments (checks in middle column).
	☐ Most of the assigned handouts/problems attempted.
	☐ Showed work and not just answers on most of the problems.
	☐ Answers to circled problems are mostly correct/accurate.
2	☐ Packet is missing more than 3 assignments (More than 3 checks in leftmost column)
	☐ More than 3 partially complete assignments.
	☐ Some of the assigned handouts/problems were attempted.
	☐ Showed work and not just answers on some of the problems.
	☐ Answers to circled problems either missing or not correct/accurate.
1	☐ Packet has only 2-3 items that may or may not be completed.
	Few or none of the handouts/assigned problems attempted.
	☐ Showed work on just a few of the problems.
	☐ Answers to circled problems are missing or mostly incorrect/inaccurate.
N	
	☐ No Evidence because packet was not turned in.
	= 110 Bytachec because packet was not tarned in

Also counts as a behavior learning target: I can manage my responsibilities as a student: BST1.2 I turn in completed work.

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Consistently/Independently (C/I)	Generally (G)	Rarely or with reminders (R)		
90% of assignments are 100% complete	Between 50% to 89% are 100% completed	More than 50% is missing or incomplete		