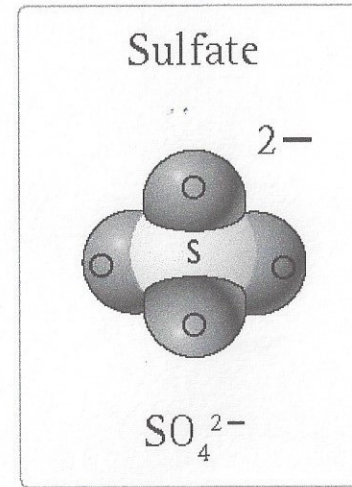
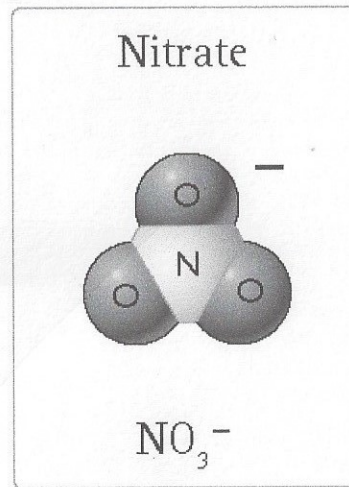
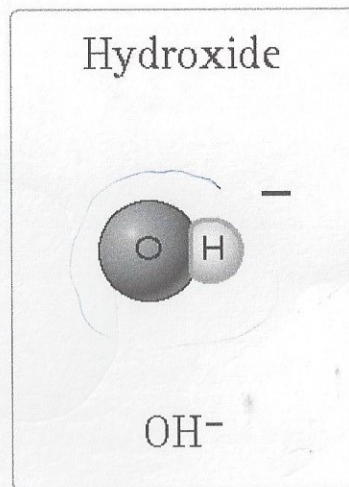
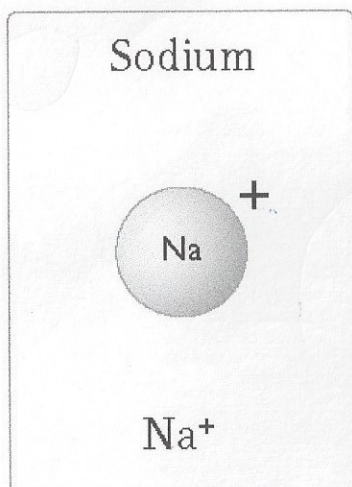


Lesson 22 ChemCatalyst

- The cards show a sodium ion and three polyatomic ions.

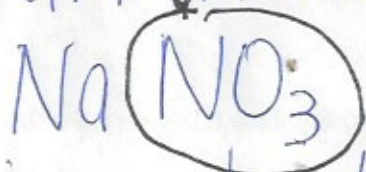


1. What do you think a polyatomic ion is?
2. Name three compounds formed between sodium ions and each of the three polyatomic ions.
3. Use the rule of zero charge to write the compounds' formulas.

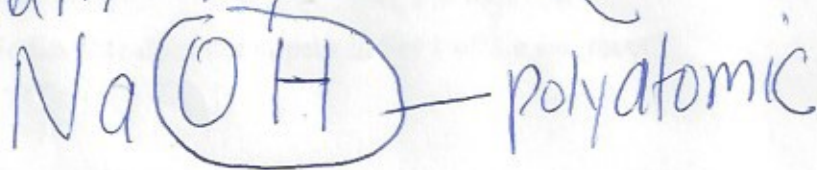
1. A polyatomic ion is one that has a lot of elements and a charge overall.

Polyatomic ion - see Warmup #2

Sodium nitrate



Sodium hydroxide



Sodium sulfate



$2+ + (2-) = 0$

