LAUNCH M, 9/21/15

Directions: Open notes to the Golden Penny Lab. Begin and entry with today's date:

What do you think happened to change the color of the penny…?

1) To silver?

2) To gold?

3) Do you think the penny you made is real gold? Why/Why not?

4) How could you find out using tools and techniques we have used in class?

Silver chemical reaction between Zinc(Zn) and Copper and sodium hydroxide

evidence - Zinc was blackish-grey

→ silver may be a coating since it rubbed off when the tongs touched.

Gold real gold is soft

gold on the penny wore off.

How could we prove it is real gold.

density - is a physical property unique to a substance and can be used to identify a substance.

$density_{gold} = 19.32 g/mL$

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Period 5

Turned silver because of the zinc and sodium hydroxide.
- silver grey zinc
- Copper atoms on the penny reacted with something (Zn) to form the silver penny.

Turned gold when we put on the hot plate.

→ heat did something to the silver penny –

silver color could be like a coating – maybe copper underneath.
- heat broken down the coating
- maybe the gold color coating is like bronze –

Cu and Zn ⇒ alloy.
(copper