

Q: How can I use mass and volume to determine the density of a substance?

• Something I like.

• margin questions (at least 1)

• Circle important vocab

• Something to check/improve

→ units!

• mass grams (g)

• liquid volume (mL) milliliters

• solid volume $\text{cm} \times \text{cm} \times \text{cm} = \text{cm}^3$
(regular)

KNOW $1 \text{ mL} = 1 \text{ cm}^3$

⇒ attempt any analysis questions.

⇒ chewing gum seems irregular

NAME NAME NAME
on all papers.